

## WEST BENGAL STATE UNIVERSITY

B.Sc. Honours 3rd Semester Examination, 2022-23

## CEMACOR07T-CHEMISTRY (CC7)

Time Allotted: 2 Hours

Full Marks: 40

The figures in the margin indicate full marks.

Candidates should answer in their own words and adhere to the word limit as practicable.

All symbols are of usual significance.

### Answer any four questions taking one from each unit

#### **UNIT-I**

1. (a) Give appropriate reagents to carry out the following transformation and explain your answer.

Me CH<sub>2</sub>

$$CH_3 - C \equiv C - CH_3$$

$$H$$

$$CH_3 - C \equiv C - CH_3$$

(b) 
$$OH \longrightarrow OH \longrightarrow OH$$
  $OH \longrightarrow OH$   $OH \longrightarrow$ 

2

3

Explain the formation of product.

(c) Carry out the following conversion:

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- (i) Z-2-butene to E-2-butene.
- (d) Write the structures of the ozonides formed when 2,3-dimethyl-2-butene was subjected to ozonolysis in the presence of HCHO. Give the mechanism for ozonide formation.

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2. (a) How can you carry out the following transformations?

2+2

(i) 
$$H_3C-C \equiv C-CH_3 \longrightarrow H_3C-CH_2-C \equiv CH$$

(b) Transform(i) CH<sub>3</sub> - CH<sub>2</sub> - C ≡ CH - CH<sub>3</sub> CH<sub>2</sub> CH<sub>2</sub> CH<sub>0</sub>

 $1\frac{1}{2}+1\frac{1}{2}$ 

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(c) What happens when 1,3-butadiene is treated with HBr at -80°C and at 40°C separately? Predict the product composition in each case and offer proper explanation in support of your answer.

# $2 \times 3 = 6$

3

#### **UNIT-II**

- 3. Answer any three from the following:
  - (a) When benzene is separately treated with Me<sub>3</sub>C-CH<sub>2</sub>Cl and Me<sub>2</sub>CH-CH(Cl)CH<sub>3</sub> in the presence of anhydrous AlCl<sub>3</sub>, same product is obtained. Identify the products with proper explanation.
  - (b) Account for the following observation: The amination of both o-chloroanisole and m-chloroanisole yields only m-anisidine.
  - (c) Acetanilide readily decolorizes bromine colour when treated with Br<sub>2</sub>/AcOH solution though it does not contain any olefinic unsaturation. Justify.
  - (d) Identify compounds A and B in the following sequence of reactions and suggest mechanism of their formation

$$\begin{array}{c|c}
Cl \\
\hline
O \\
Br
\end{array}
\xrightarrow{Mg} A \xrightarrow{CH_3CHO} B$$

4. Answer any three from the following:  $2 \times 3 = 6$ 

(a) Write down the product of the following reaction with mechanism.

$$\begin{array}{c|c}
OH & & POCl_3 / DMF \\
\hline
OH & & H_3O^+
\end{array}$$
?

- (b) Mention the proper position of E<sup>(+)</sup> in the product of the following reactions
  - $CH = CH COOH \xrightarrow{E^{(+)}}$
  - $O \rightarrow B(OH)_2 \xrightarrow{E^{(+)}}$
- (c) Suggest the most suitable method for the following conversion.

(d) Rationalize the fact with suitable mechanism that nitration of 4-tbutyl toluene gives 4-nitro toluene as one of the products.

#### **UNIT-III**

5. (a) Carry out the following transformation and explain.

(ii) 
$$\stackrel{\text{OH}}{\swarrow}_{\text{CHO}}$$
  $\stackrel{\text{CHO}}{\longrightarrow}$   $\stackrel{\text{CH}}{\searrow}_{\text{CH}}$ 

2+2

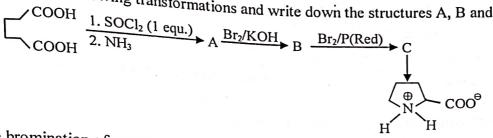
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(b) Predict product(s) for the following reaction

PhCHO 
$$\frac{D_2O}{50\% \text{ NaOH}}$$
?

PhCDO  $\frac{50\% \text{ NaOH}}{H_2O}$ ?

(c) Complete the following transformations and write down the structures A, B and C.



- (d) The bromination of acetone is catalysed by acids and it is zero order with respect to
- (e) Define atom economy. Give example. 2
- (f) What product(s) are obtained when benzaldehyde is treated with propanoic 2 anhydride and sodium propanoate? Give mechanism of the reaction.
- (g) Suggest a suitable mechanism for the following transformation. 2

- 6. (a) Predict the product in the following with plausible explanation.

  - (b) The compound (A) in the following, on hydrolysis, yields (±) 1-phenylethanol. 2 Explain.

(c) Identify the products in the following reaction and explain

(d) Carry out the following conversions mentioning proper reaction conditions and 2+2reagents along with plausible mechanism.



(ii) Acetophenone → phenylvinyl ketone

 $1\frac{1}{2}+1\frac{1}{2}$ 

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2

2

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- (e) Predict the major product of the following reaction and explain why it is major one (any one)
- 3

(i) i. 
$$\frac{1}{N}$$
ii. PhCH<sub>2</sub>Cl
iii. H<sub>3</sub>O<sup>+</sup>

- (ii)  $\bigcirc$  OH  $\longrightarrow$  CHO  $\longrightarrow$  CHO  $\longrightarrow$  ?
- (f) Explain the following statements.

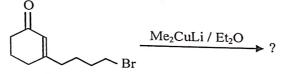
2+2

- Semicarbazide hydrochloride does not react with a ketone alone unless sodium acetate is mixed.
- (ii) Chloral is obtained in hydrated form only. Explain.

#### **UNIT-IV**

7. (a) Give the products with proper explanations:

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(b) Predict the product with plausible mechanism in each of the following reactions:

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(c) How would you synthesize the following compound with the help of Reformatsky reaction?

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$$PhC(Me) = C(Me) COOH.$$

8. (a) What happens when diisopropyl ketone is allowed to react with  $(i-Pr)_2MgBr$ ? Give mechanism. Do you expect the same product if diisopropyl ketone is allowed to react with  $(i-Pr)_2CHLi$ ?

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(b) On treatment with Mg in dry ether, allyl-bromide gives hexa-1,5-diene whereas propyl bromide forms corresponding Grignard reagent — Justify.

 $1\frac{1}{2}$ 

(c) Carry out the following transformation.

 $1\frac{1}{2}$ 

Acetone 
$$\longrightarrow$$
 Me<sub>2</sub>C  $-$  CH<sub>2</sub>OH  $\stackrel{|}{\circ}$  OH

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